Surname		Centre Number	Candidate Number
Other Names			2
wjec cbac	GCE AS – NEW AS B400U10-1 BIOLOGY – Component 1 Basic Biochemistry and Cell Organ	P	eduqas Part of WJEC

P.M. THURSDAY, 26 May 2016

1 hour 30 minutes

For Examiner's use only						
Question	Maximum Mark	Mark Awarded				
1.	9					
2.	6					
3.	7					
4.	16					
5.	13					
6.	15					
7.	9					
Total	75					

### ADDITIONAL MATERIALS

In addition to this examination paper, you will need a calculator and a ruler.

#### INSTRUCTIONS TO CANDIDATES

Use black ink or black ball-point pen. Do not use gel pen. Do not use correction fluid.

Write your name, centre number and candidate number in the spaces at the top of this page.

Answer **all** questions.

Write your answers in the spaces provided in this booklet. If you run out of space, use the continuation pages at the back of the booklet, taking care to number the question(s) correctly.

#### INFORMATION FOR CANDIDATES

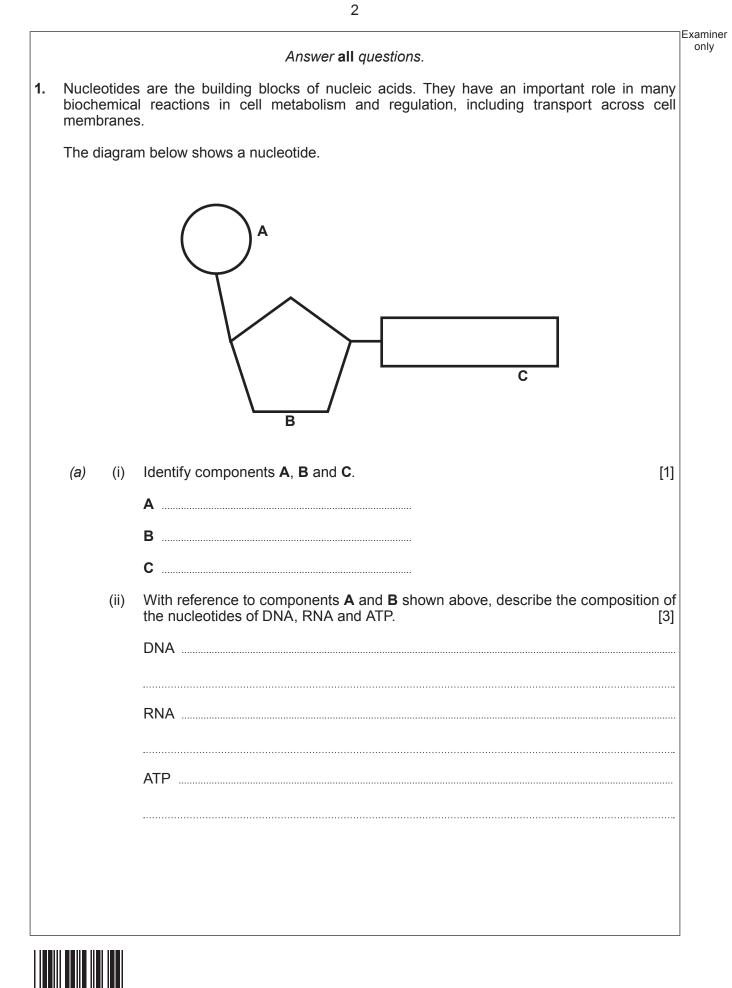
The number of marks is given in brackets at the end of each question or part-question.

The assessment of the quality of extended response (QER) will take place in question 7.

The quality of written communication will affect the awarding of marks.



B400U101 01



(b) An experiment was carried out to investigate the rates of uptake of different sugars by the small intestine. One experiment used normal intestine, the second used a piece of intestine treated with cyanide. The results are shown in the table below.

	Relative rates of absorption/a.u.				
Sugar	Normal intestine	Intestine treated with cyanide			
glucose	1.00	0.33			
galactose	1.10	0.53			
xylose	0.30	0.31			
arabinose	0.29	0.29			

- (i) Name **two** sugars from the table which can be absorbed by active transport. [1]
- (ii) Using evidence from the table, explain why you chose these sugars.

It was concluded that all of the sugars named in the table can be absorbed by diffusion. (C) Under what conditions could this take place and how does evidence from the table support this conclusion? [2]



Turn over.

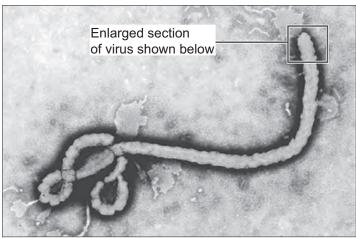
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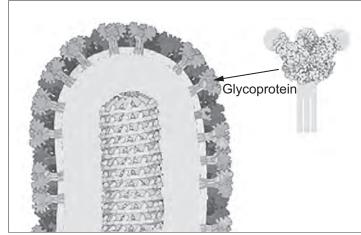
B400U101 03

[2]

Examiner only 2. Ebola virus disease is a serious illness of humans that originated in Africa, where there was an outbreak which started in 2014. The photographs and diagrams below show the Ebola virus and an enlarged section of the virus in detail.



Ebola virus



Enlarged section of Ebola virus

In addition to the usual viral structure, Ebola is surrounded by a lipid-bilayer, which is derived from infected cell membranes as the virus buds from the cell.

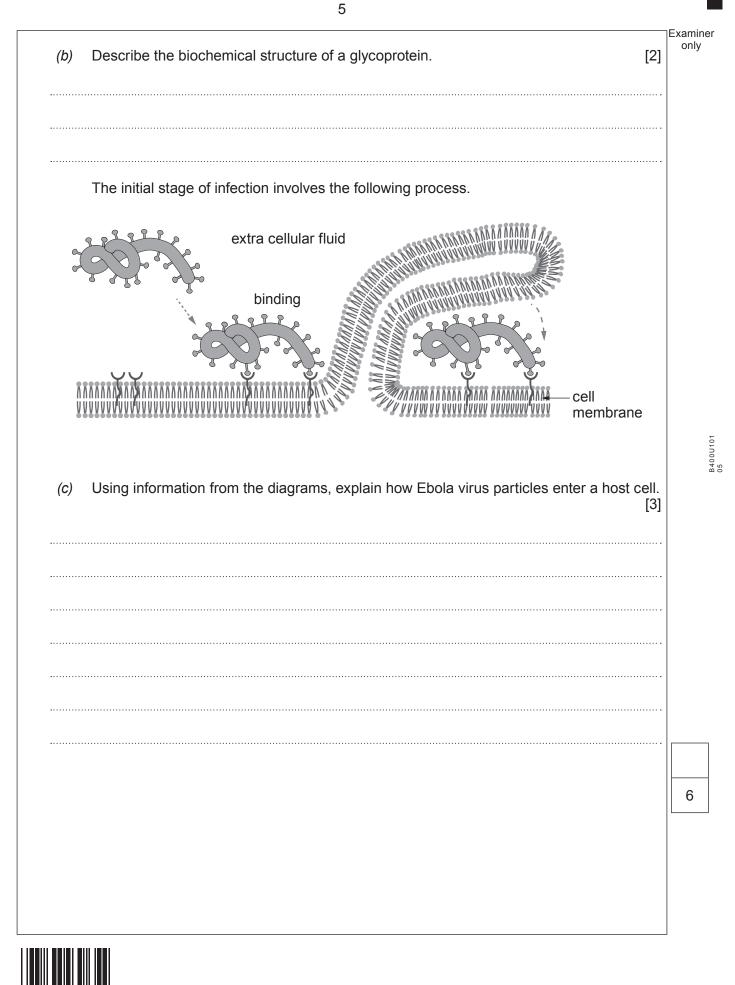
A viral transmembrane glycoprotein, is incorporated into this membrane and allows the virus to bind to blood vessel cells.

(a) What are the **two** major biochemical components present in **all** viruses?

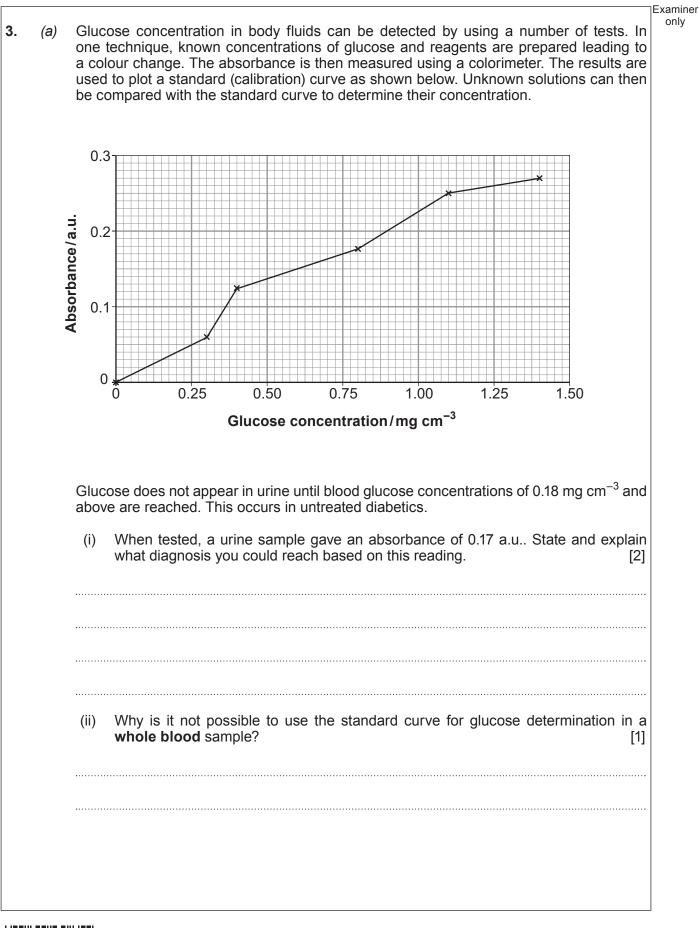
[1]

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Examiner only (b) Modern medicine uses immobilised enzymes, in devices called biosensors, to detect blood glucose levels. This involves the use of glucose oxidase as the enzyme and glucose as the substrate. Give two advantages of using immobilised enzymes for blood glucose monitoring. [2] (C) The following structures show two monosaccharides. CH<sub>2</sub>OH CH<sub>2</sub>OH Н OH HO С OH С Н Н С С С С OH OH Н Н Ċ HO Ċ Н Н С С н Н OH Н OH Glucose Galactose Describe the difference between these two molecules. [1] (i) (ii) What would be the products of a condensation reaction between these two molecules? [1] 7

7



B400U101 07

An experiment was carried out to determine the change in mass of potato tissue immersed in different sucrose solutions. Skinless potato cylinders were prepared in order to carry out this experiment.
(a) Calculate the total surface area, to one decimal place, of one of the cylinders which had a length of 45 mm and a diameter of 8 mm. [3]
Surface area of a cylinder is 2 π r<sup>2</sup> + 2 π r h r = radius h = length π = 3.14

Surface area = ..... mm<sup>2</sup>

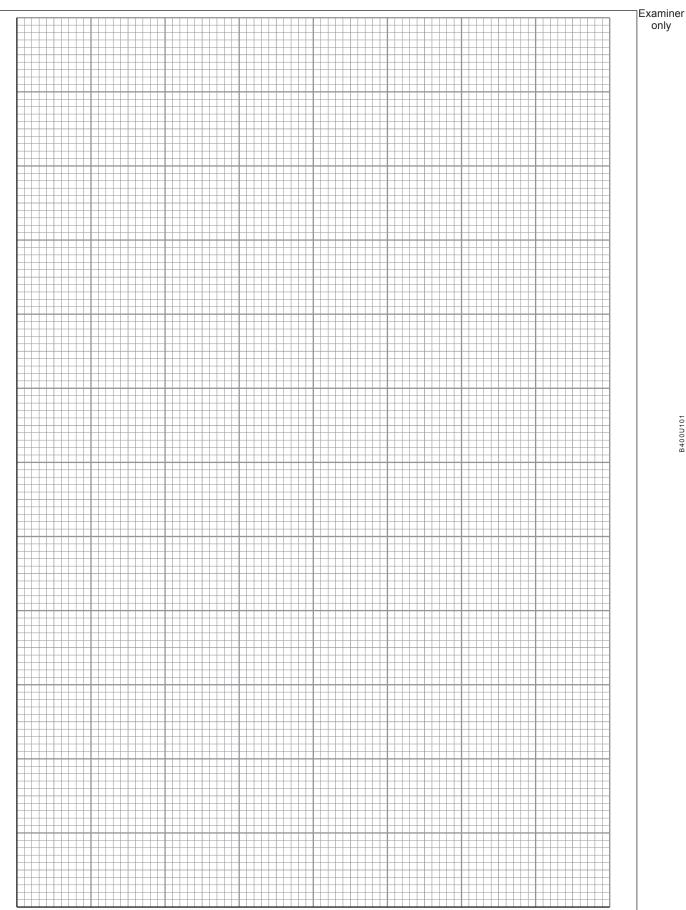
Sucrose solutions were prepared at different concentrations. The skinless potato cylinders of equal length from the same potato were weighed and each immersed in one of the solutions. After two hours they were blotted dry and reweighed. The percentage change in mass was calculated.

The results are shown below.

Sucrose Concentration /M	Initial Mass /g	Final Mass /g	Change in Mass /g	Percentage Change in Mass/%
0.0	3.4	4.1	0.7	20.6
0.2	3.4	3.5	0.1	2.9
0.4	6.2	5.3	-0.9	-14.5
0.6	6.3	4.8	-1.5	-23.8
0.8	6.2	4.4	-1.8	-29.0
1.0	6.0	4.3	-1.7	-28.3

(b) Use the data above to **draw a graph** on the page opposite showing how percentage change in mass of potato is affected by the change in sucrose concentration. [3]







Sucrose concentration /M	Water potential /kPa	
0.05	-130	
0.10	-260	
0.15	-410	
0.20	-540	
0.25	-680	
0.30	-860	
0.35	-970	
0.40	-1120	
0.45	-1280	
0.50	-1450	
0.55	-1620	
0.60	-1800	
0.65	-1980	
0.70	-2180	



(c)Using the conversion table below, determine the water potential of the potato tissue in this

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only Explain what has caused the percentage difference in the mass of the tissue in the following concentrations of sucrose solution in terms of water potential. [4] 0.8 M The experiment was repeated using the same range of sucrose concentrations and sweet potato but there was no decrease in mass at any sucrose concentration. Explain why there was no decrease in mass at any of the sucrose concentrations. [1] Describe how you would modify the method to determine the  $\boldsymbol{\psi}_{\text{tissue}}$  of sweet [2]



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(d)

(e)

(i)

(ii)

potato.

0.0 M

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An experiment was set up as shown below with the addition of alkaline sodium carbonate causing 5. the milk to turn pink. Phenolphthalein is a pH indicator which is pink in alkaline conditions and colourless when in acid conditions. The experiment was timed to see how long it took for the indicator to turn from pink to colourless. stirring rod stirring rod 1 cm<sup>3</sup> lipase solution add in turn: 5 cm<sup>3</sup> milk 7 cm<sup>3</sup> sodium carbonate add in turn: 5 cm<sup>3</sup> milk 7 cm<sup>3</sup> sodium carbonate solution solution 5 drops of phenolphthalein 5 drops of phenolphthalein stir and start timing when you add the lipase Explain what caused the indicator to lose the pink colour and become colourless. [3] (a) Why would this experiment not be suitable for assessing the effect of different pH values (b) on lipase? [1]

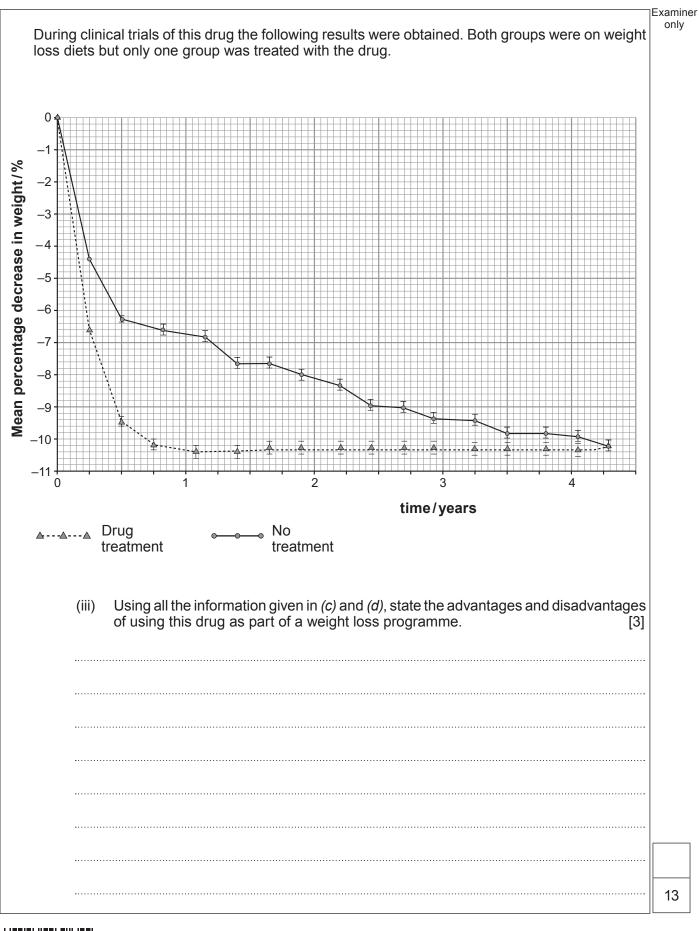
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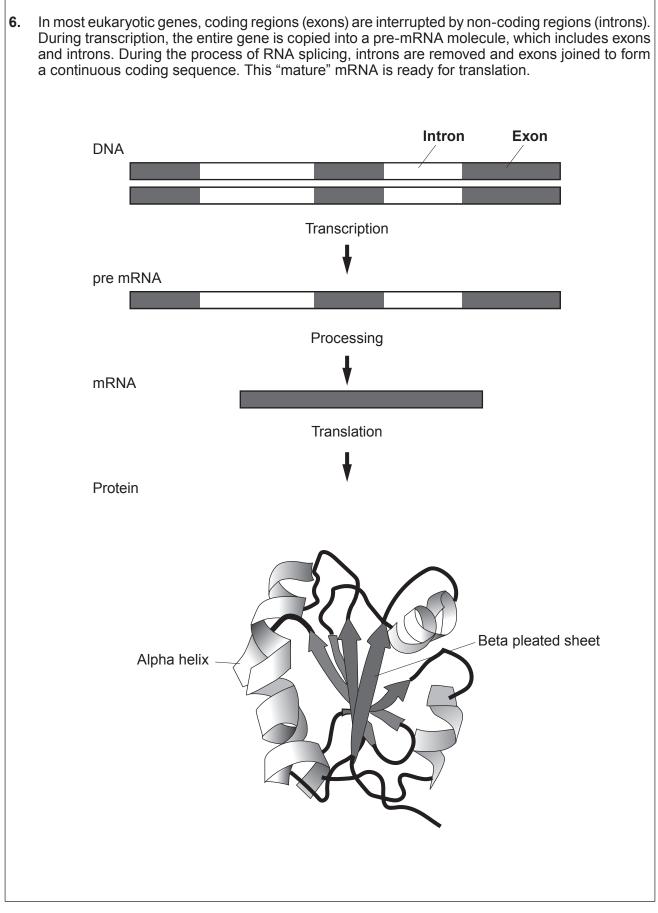
Weight loss in obese patients can be achieved by a reduction in their intake of lipids or by reducing their ability to digest and / or absorb lipids. One drug used to treat obesity acts as a competitive inhibitor of lipase. It prevents approximately 30% of the lipid being absorbed, thereby reducing energy (calorie) intake. The extra lipid in the large intestine can lead to unpleasant side effects such as diarrhoea and deficiency diseases. Treatment of patients requires them to take vitamin tablets since absorption of fat-soluble vitamins and other fat-soluble nutrients is inhibited by the use of the drug. After drug treatment was stopped, a significant number of subjects regained up to 35% of the weight they had lost. Describe and explain the action of this drug on pancreatic lipase. [3] (C) (d) Describe how this drug would lead to weight loss. [2] (i) With reference to the action of this drug, explain why people tended to regain the (ii) weight when they stopped taking it. [1]







Turn over.





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(a)	Expl	ain the process of transcription including the roles of the enzymes involved.	[3]
			·····•
′b)	(i)	Explain why it is important that the RNA which is finally translated, consists onl exons.	y of [2]
	(ii)	Explain how a change in the sequence of bases in the DNA of the gene (a mutat would affect the protein produced if it was in	ion)
		I. an intron;	[1]
			·····
		II. an exon.	[1]



Normal	Т	А	С	А	А	A	G	т с	А	С	С	А	С	Т	template strand
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			First Base	U	U Phe Leu Leu Leu Leu Leu	1 1 1 1 1 1 1 1 2 5	Seco Ser Ser Ser Pro Pro Pro Pro Thr	nd Base A Tyr Tyr Stop Stop Hls Hls Gln Gln Asn		G Cys Stop Trp Arg Arg Arg Arg Ser	U C A G U C A G U	Third Base			
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			First Base	U C A	U Phe Leu Leu Leu Leu Leu Ile Ile	e e 1 1 1 1 1 1 1 1 1	Seco Ser Ser Ser Pro Pro Pro Pro Thr Thr Thr	nd Base A Tyr Tyr Stop Stop Hls Hls Gln Gln Gln Asn Asn Lys Lys		G Cys Cys Cys Cys Cys Cop Trp Arg Arg Arg Ser Ser Arg Arg Arg	U C A G U C A G U C A G	Third Base			
			First Base	U C	U Phe Leu Leu Leu Leu Leu Ile Ile Ile Val	e e J J J J J J J L L	Seco Ser Ser Ser Pro Pro Pro Pro Thr Thr Thr Thr Ala	nd Base A Tyr Tyr Stop Stop His His Gin Gin Gin Asn Lys Lys Asp		G Cys Cys Stop Trp Arg Arg Arg Ser Arg Arg Ser Arg Gly	U C A G U C A G U C A G U U C	Third Base			



(ii)	Explain why the change of bases shown is described as a silent mutation. [1]	Examiner only

Chargaff carried out work examining the ratios of different bases in samples of DNA from different sources. A sample of his results is shown below.

### Percentage composition

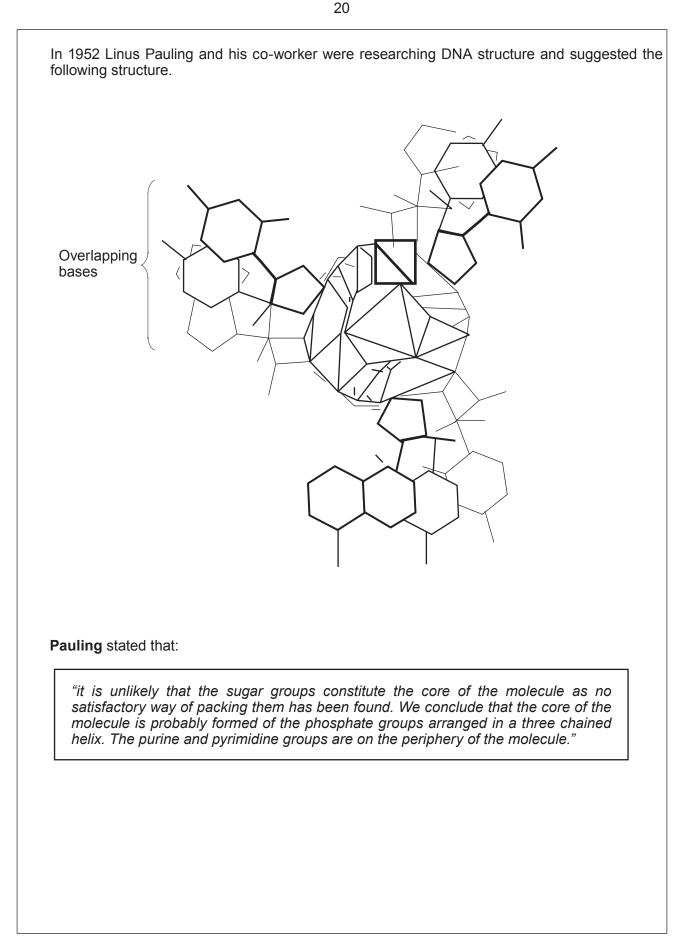
Source of DNA	Adenine	Thymine	Guanine	Cytosine
Yeast	31.3	32.9	18.7	17.1
Herring sperm	27.8	27.5	22.1	22.6
Human sperm	30.7	31.2	19.3	18.8

Calculate the exact ratio to two decimal places of adenine to thymine and guanine (d) (i) to cytosine for human sperm. [2]

adenine to thymine ratio =

guanine to cytosine ratio =

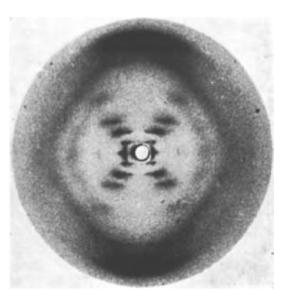






At this time, Rosalind Franklin was carrying out X-Ray crystallography on pure DNA samples.

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X-ray diffraction image of the double helix structure of the DNA molecule, taken in 1952 by Raymond Gosling, commonly referred to as "Photo 51", during work by Rosalind Franklin on the structure of DNA.

In 1952 Franklin concluded that:

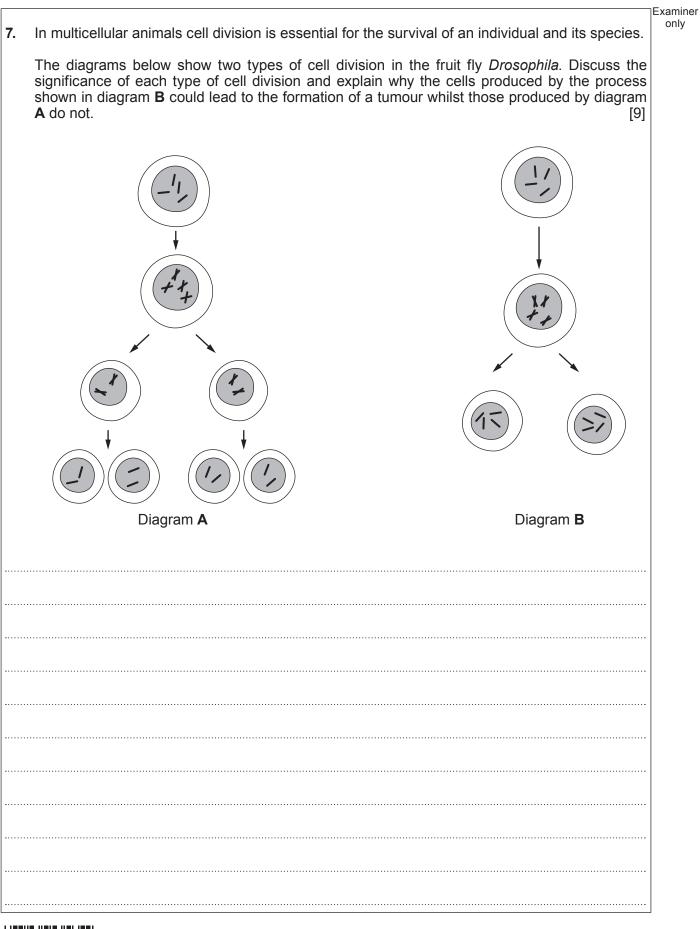
"The results suggest a helical structure of DNA, containing probably 2, 3, or 4 nucleic acid chains and having the phosphate groups near the outside."

(ii) Evaluate how closely the suggestions by Pauling and Franklin, shown in the boxes above and opposite, match the current understanding of the arrangement of nucleotides in DNA.

Pauling	
Franklin	
	 15



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